



## Baseline

## Distribution of surfactants along the estuarine area of Selangor River, Malaysia



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## ABSTRACT

This study aims to determine the levels of methylene blue active substances (MBAS) and ethyl violet active substances (EVAS) as anionic surfactants and of disulphine blue active substances (DBAS) as cationic surfactants in the surface microlayer (SML) around an estuarine area using colorimetric methods. The results show that the concentrations of surfactants around the estuarine area were dominated by anionic surfactants (MBAS and EVAS) with average concentrations of 0.39 and 0.51  $\mu\text{mol L}^{-1}$ , respectively. There were significant between-station differences in surfactant concentrations ( $p < 0.05$ ) with higher concentrations found at the stations near the sea. The concentration of surfactants was higher during the rainy season than the dry season due to the influence of runoff water. Further investigation using total organic carbon (TOC) and total organic nitrogen (TON) shows that there is a significant correlation ( $p < 0.05$ ) between both anionic and cationic surfactants and the TON concentration.

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Estuaries constitute transition zones or ecotones where fresh water from land drainage mixes with seawater, creating some of the most biologically productive areas (Kennish, 2002). The estuary ecosystem is significantly important in terms of biogeochemical cycles due to high levels of nutrient and organic substances which inflow to the ecosystem (Turner and Millward, 2002; Martinho et al., 2008; Shan et al., 2013). The surface microlayer (SML) in estuarine areas is a specific ecosystem due the influence of river water intrusion. The SML is a layer within  $50 \pm 10 \mu\text{m}$  that is generally recognised by high levels of particulates and dissolved organic and inorganic compounds (Žutić et al., 1981; Zhang et al., 2006; Cunliffe et al., 2013). Several pollution sources of organic matter in the SML include plant and animal secretions, bacterial decomposition, autolysis processes, fertilizer, sewage, effluents, and erosion (Bodineau et al., 1999; Bianchi, 2001; Volkman and Tanoue, 2002; Zhang et al., 2008). Previous research shows that there are three major classes of biochemical compounds in the SML: carbohydrates, proteins and lipids (Sieburth et al., 1976; Williams et al., 1986; Calace et al., 2007).

Organic compounds with surfactant properties are an important group of organic substances in the SML due to their ability

to increase the solubility of inorganic and organic substances (Li and Chen, 2002; Doong and Lei, 2003; Castro et al., 2005). The largest proportion of naturally occurring surfactants originates from phytoplankton exudates in sub-surface water (SSW); a water layer between 10 cm and 1 m from the water's surface (Pikkariainen and Lemponen, 2005; Nakajima et al., 2013). SSW comprise numerous lipids, proteins and their degradation products, glycopeptides–lipid–oligosaccharide complexes and pigments accumulating at the marine interface (Vojvodić and Čosović, 1996; Pavlič et al., 2005; Wurl et al., 2011). Surfactants in the SML can also be generated photo-chemically from the degradation of refractory dissolved organic matter (DOM), such as humic substances (Kieber and Mopper, 1987; Kieber et al., 1990; Mopper et al., 1991). Photo-oxidation and degradation processes of humic substances containing higher aromatic and carbohydrate content by microorganisms produce smaller material organic materials dominated by aliphatic components with carboxyl functionalities (Abdulla et al., 2013). This would result in the addition of surfactants with carboxylic acid functional groups (McKnight and Aiken, 1998). In coastal areas, surfactant concentrations may be generated due to urbanisation and the contribution of waste and runoff water (Zhang et al., 2006; Stortini et al., 2009).

The amount of surfactants in the ocean's surface can determine the fate of chemicals in the SML and contribute to the amount of

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