

## Effect of Ethanol and Methanol on the Total Phenolic Content and Antioxidant Capacity of Chia Seeds (*Salvia hispanica* L.)

(Kesan Etanol dan Metanol ke atas Jumlah Kandungan Fenolik dan Kemampuan Antioksidan dalam Biji Chia (*Salvia hispanica* L.))

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### ABSTRACT

*The use of methanol and ethanol for the extraction of antioxidant compounds in Chia Seed (*Salvia hispanica* L.) is studied, but only the effect of a single concentration of these solvents was evaluated. However, factors such as the nature and the concentration of the solvents determines the content of phenolic compounds, the antioxidant capacity and the possible use of the extract. For this reason, it is necessary to have a detailed study of different extraction conditions to find the optimum. The objective of this work was to evaluate the effect of different concentrations of ethanol and methanol and the hexane treatment on the extraction of phenolic compounds and the antioxidant capacity of the Chia Seeds extracts. The greatest total phenolic content was obtained in the 50% aqueous ethanol and methanol extracts. Nevertheless, all the extracts showed a high DPPH scavenging activity. With exception of 100% ethanol extract, no significant differences were found between the extracts treated with and without hexane. No significant differences were found in the extraction yields of the 80% aqueous ethanol and methanol extracts, but the aqueous methanol extract presented lower  $IC_{50}$ . The FTIR and UV-VIS spectra of the aqueous ethanol and methanol Chia extracts suggest differences in the composition of the extracts. It is possible to extract the phenolic compounds from the Chia Seeds with different concentration of ethanol and methanol, which will allow to select the appropriate extraction conditions according to the application of the extract.*

*Keywords: Antioxidants; Chia Seeds; hexane treatment; phenolic compound; solvent extraction*

### ABSTRAK

*Penggunaan metanol dan etanol untuk pengekstrakan sebatian antioksidan dalam Biji Chia (*Salvia hispanica* L.) telah dikaji, tetapi hanya kesan satu kepekatan pelarut telah dinilai. Faktor seperti sifat dan kepekatan pelarut menentukan kandungan sebatian fenolik, keupayaan pengantioksidan ekstrak serta kegunaan ekstrak. Oleh itu, kajian terperinci tentang keadaan pengekstrakan yang berbeza diperlukan untuk mendapatkan nilai yang optimum. Objektif kajian ini adalah untuk menilai kesan kepekatan etanol dan metanol yang berbeza dan rawatan heksana terhadap pengekstrakan sebatian fenolik dan keupayaan antioksidan ekstrak Biji Chia. Kandungan fenolik terbesar diekstrak menggunakan 50% larutan etanol dan metanol. Walau bagaimanapun, semua ekstrak menunjukkan aktiviti hapus sisa DPPH yang tinggi. Kecuali ekstrak etanol 100%, tiada perbezaan yang ketara dijumpai antara ekstrak-ekstrak yang dirawat dengan heksana dan tanpa heksana. Tidak terdapat perbezaan yang ketara dalam hasil pengekstrakan 80% etanol dengan ekstrak metanol, tetapi ekstrak larutan metanol menunjukkan  $IC_{50}$  yang lebih rendah. Spektrum FTIR dan UV-VIS daripada ekstrak larutan etanol dan ekstrak metanol Biji Chia menunjukkan perbezaan dalam komposisi ekstrak. Sebatian fenol boleh diekstrak daripada Biji Chia dengan kepekatan etanol dan metanol yang berbeza, yang boleh menentukan kegunaan ekstrak mengikut kesesuaiannya.*

*Kata kunci: Biji Chia; pengantioksidan; pengekstrakan pelarut; rawatan heksana; sebatian fenol*

### INTRODUCTION

Chia Seeds (*Salvia hispanica* L.) have acquired great value for society due to their nutritional and medicinal properties. As a food, they contribute a significant amount of proteins (15-25%), carbohydrates (26-41%), fatty acids (30-33%) and fibre (18-30%) (Ixtaina et al. 2008).

In medicine, has been described that they protect against degenerative diseases such as atherosclerosis, cancer, Alzheimer's and Parkinson's (Corona et al. 2016; Valdivia & Tecante 2015). These properties are attributed to the phenolic compounds with potential antioxidant activity that they contain, such as the myricetin, quercetin, and kaempferol (Capitani et al. 2012; Reyes et al. 2008). The

total phenolic content reported in Chia Seeds range from 0.64 to 1.63 mg GAE/g in 100% ethanol and 70% aqueous methanol extracts (Capitani et al. 2014; Martínez-Cruz & Paredes-López 2014; Silveira & Salas 2014). The phenolic content in Chia Seed is higher than in other plant species (Kähkönen et al. 1999). Antioxidants are also used as additives in food preservation and, due to the importance of these molecules, many investigations seek methodologies that allow their extraction with high yields. One of the techniques widely used is the extraction with solvents, such as water, ethanol, methanol, hexane among others (Santos & Gonçalves, 2016). However, the nature of the solvent, their concentration and the time of exposure, are factors that affect the efficiency of the extraction of the phenolic compounds (Azmir et al. 2013). And they delimit the application of the extracts, there are restrictions on the use of certain solvents, due to their toxicity according to different legislations. In Chia Seeds, it is not possible to use only water for the extraction of phenolic compound. When the seed is mixed with the water, releases a tetra saccharide called mucilage (Muñoz et al. 2012). The mucilage retains water and forms a highly viscous solution that avoids the extraction of phenolic compounds. For this reason, it is necessary to use other polar solvents and their mixtures with water. Methanol is known as an efficient agent for its extraction (Do et al. 2014) but it is toxic (Villanueva et al. 2002). On the other hand, ethanol represents a safe solvent alternative in separation processes of compounds for human consume (Ignat et al. 2011). In some methods, it is proposed that before the extraction of phenolic compounds, the lipids are removed from the samples or washed the extracts with hexane, in order to obtain purer extracts and quantify a higher content of phenolic compounds (Silveira & Salas 2014). However, the use of another solvent, makes the process more expensive and limits the application of the isolate because hexane is considered toxic (Buddrick et al. 2013). The objective of this work was to evaluate the effect of different concentrations of ethanol and methanol and the hexane treatment on the extraction of phenolic compounds and the antioxidant capacity of the Chia Seeds extracts. Likewise, techniques such as UV-Visible and FTIR were applied to analyse the extracts. This study will establish the adequate conditions for the extraction of antioxidant compounds from the Chia Seeds and their potential applications.

## MATERIALS AND METHODS

### SEEDS AND REAGENTS

Chia Seed was purchased from a local market in Puebla City, Mexico. They were cleaned manually and were passed through a 20-mesh screen. The flour was generated by grinding the seeds in a food processor and passing

them through a 35-mesh screen. The flour was stored at 4 °C until its use. The 2,2-diphenyl-1-picrylhydrazil (DPPH), Folin-Ciocalteu reagent, Gallic acid and standards HPLC grade (Quercetin  $\geq$  95%, Kaempferol  $\geq$  90%, Myricetin  $\geq$  96.0%, Caffeic acid  $\geq$  98.0%, Trans-cinnamic acid  $\geq$  99.0%) were obtained from Sigma Aldrich (Sigma Co., San Luis, U.S.A.).

### EXTRACTION AND DETERMINATION OF TOTAL PHENOL CONTENT

The phenolic compounds were extracted using the methods proposed by Martínez-Cruz and Paredes-López (2014). 0.5 g of chia flour was mixed with 3 mL of different concentrations (50, 60, 70, 80, 90%) of ethanol and methanol in water and 100% ethanol and methanol, respectively. The samples were kept under agitation for 24 h at 25 °C and were centrifuged at 6,000 rpm for 10 min. For hexane treatment of the extracts: 3 mL of hexane was added to the ethanol and methanol extracts, stirred for 5 min at 25 °C and the phases were separated with a separatory funnel. This procedure was performed three times. The extracts were stored at 10 °C until analysis. A completely randomized design was carried out with a  $2 \times 2 \times 6$  factorial arrangements. The quantification of total phenolic content was determined using the Folin-Ciocalteu method (Swain & Hillis 1959) in microplate. The gallic acid was used as standard in the calibration curve, the results were expressed in milligrams equivalents of gallic acid per gram of Chia Seed (mg GAE/g of Chia Seed). The results show the mean of three replicates  $\pm$  standard deviation (SD).

### DPPH RADICAL SCAVENGING ACTIVITY

The antioxidant capacity of the extracts was determined by DPPH assay (Corral et al. 2008) in microplate. 280  $\mu$ L of the 100 mM DPPH solution was mixed with 20  $\mu$ L of each of the extracts. The percentage of the DPPH scavenging activity was determined using (1):

DPPH scavenging activity (%) =

$$\frac{Absorbance_{control} - Absorbance_{sample}}{Absorbance_{control}} \times 100\% \quad (1)$$

The results show the mean of three replicates  $\pm$  SD.

### IC<sub>50</sub> AND EXTRACTION YIELD

The alcohol of extracts was removed by rotary evaporator and the water by lyophilisation. From the lyophilised, solutions were prepared with a concentration of 1-10 mg/mL in water. The half-maximal inhibitory concentration (IC<sub>50</sub>) was reported as the amount of antioxidant required

to decrease the initial DPPH concentration by 50%. The extraction yield was calculated considering the initial grams of seeds, from (2):

$$\text{Extraction yield (\%)} = \text{g of extract} / \text{g of seeds} \times 100\% \quad (2)$$

#### SPECTRAL SCANNING BY UV-VISIBLE AND STRUCTURAL ANALYSIS BY FTIR

The 80% aqueous ethanol and methanol extracts without and with hexane treatment and the standards were dissolved in water. The UV-Visible scan was performed in the wavelength range of 200-900 nm. The lyophilized extracts were characterized with a FTIR spectrometer (Bruker Vertex 70, Germany) equipped with a total attenuated reflectance (ATR) accessory. The spectral measurements were recorded in the wavenumber range between 4000-500  $\text{cm}^{-1}$ .

#### STATISTICAL ANALYSIS

Analysis of variance (ANOVA) and Tukey tests were performed using the Minitab 16.0 program (Minitab, Inc., State College, PA, USA). The significant differences between the values of the means for each treatment were calculated with a probability of  $P < 0.05$ .

#### RESULTS AND DISCUSSION

##### TOTAL PHENOLIC CONTENT

Figure 1 shows the total phenolic content of the extracts obtained from the calibration curve with gallic acid (0-100  $\mu\text{g/mL}$ ,  $y = 0.0017x + 0.0519$ ,  $R^2 = 0.99$ ). In both series of ethanol and methanol extracts a similar trend was obtained. As the alcohol concentration decreases and

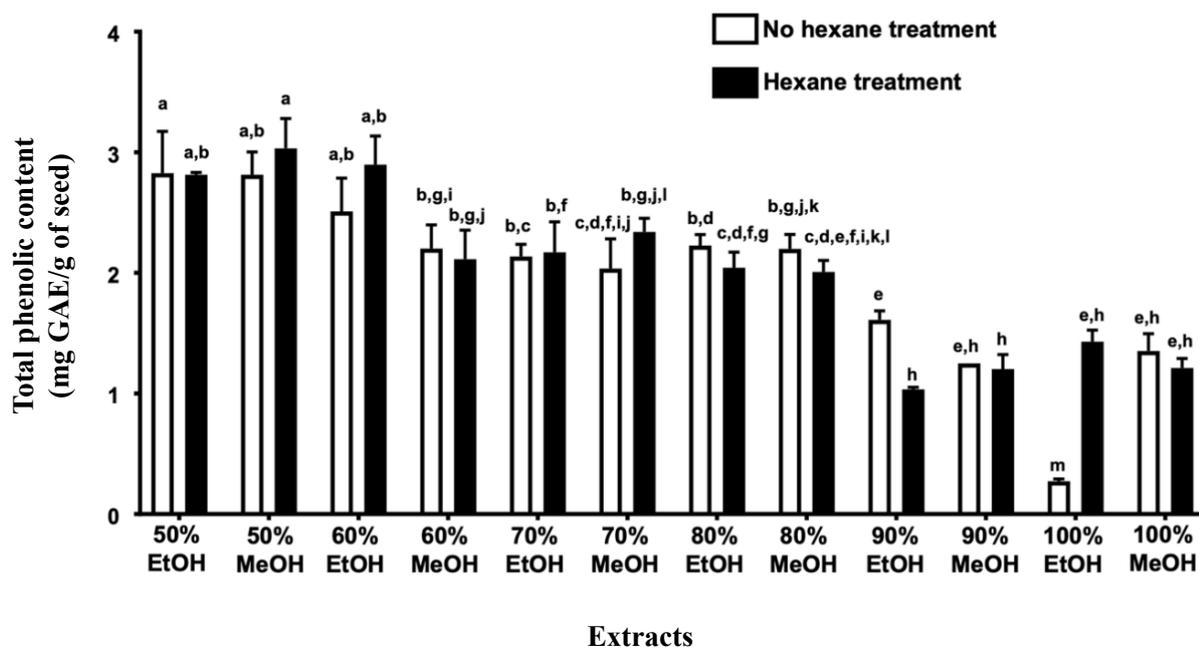


FIGURE 1. Total phenolic content of ethanol (EtOH) and methanol (MeOH) extracts of *Salvia hispanica* L. seeds with and without hexane treatment. Columns with different letter are statistically different according to the Tukey test ( $P < 0.05$ )

the content of water increases, the content of phenolic compounds increases. Concentrations  $< 50\%$  were not considered because the release of the mucilage from the Chia Seed is favoured. The mucilage retains water

and generates a viscous solution that does not permit the extraction of phenolic compounds.

The highest phenolic content in the ethanol extracts without and with treatment was quantified in the 50 and

60% aqueous ethanol extracts. However, no significant differences were found between the total phenolic content in the 60, 70, and 80% extracts. The lowest phenolic content was obtained in the 90, and 100% ethanol extracts. On the other hand, with exception of 100% ethanol extract, no significant differences were found between the phenolic content of the ethanol extracts without and with hexane treatment. After hexane treatment, in the 100% ethanol extract a greater content of compounds was quantified. This may suggest that compounds of different nature such as terpenes, saponins and others may be extracted with 100% ethanol. These molecules can interact with the phenols forming complex structures that interfere with the quantification of the total phenols content (Do et al. 2014). The results demonstrate that, hexane washes could eliminate these compounds from the extracts, so that after treatment a greater content of phenols is quantified.

The total phenolic compounds extracted with the aqueous ethanol solutions was higher than that reported in the literature. Reyes et al. (2008) quantified 0.88 mg GAE/g of chia in the 100% ethanol extract obtained from the mixture of the defatted flour with the solvent in relation 1:10, under stirring for 48 h. Marineli et al. (2014) with the same conditions reported  $0.94 \pm 0.0$  mg GAE/g of Chia. In this study, in the 100% ethanol extract generated by mixing the flour with ethanol in relation 1:6 and treated with hexane, 36% more phenolic compounds were extracted. The highest phenolic content in methanol extracts without and with hexane treatment was obtained in the 50% aqueous methanol extract. No significant differences were found between 60, 70, and 80% aqueous methanol extracts. And as in ethanol extracts, the lowest concentration of phenols was quantified in the 90%, and 100% methanol extracts. Hexane treatment did not had effect on the content of phenolic compounds in the methanol extracts. No significant differences were found between methanol extracts without and with treatment. The total phenolic content extracted with 100% methanol was 50% more than that reported by Silveira and Salas (2014). They reported 0.64 mg GAE/g of chia in the 100% methanol extract obtained of mixing the flour with the solvent in a ratio of 1:8 and treated with hexane. No significant differences were found between total phenolic content reported by Martínez-Cruz and Paredes-López (2014) in the 70% aqueous methanol extract without hexane washes and the obtained in this investigation under the same conditions. Among the ethanol and methanol extracts without hexane treatment, significant differences were found in the total phenolic content of the 100% ethanol extracts. The phenolic compounds in the 100% methanol extract was significantly higher than that of the 100% ethanol extract. Finally, there were no

significant differences between ethanol and methanol extracts treated with hexane. It has been reported that the extraction of phenolic compounds is more efficient with methanol (Castañeda et al. 2009; Ignat et al. 2011) and in this research similar results were obtained between ethanol and methanol when mixed with water; however, 100% methanol extracted higher amount than 100% ethanol (Figure 1). In this study, it was observed that as water concentration increase, the amount of phenolic content increase, and the behaviour in both alcohols is the same. This indicate that the water favours the extraction of the phenolic compounds in Chia Seeds. The solubility of antioxidant compounds in a solvent depends on the characteristics of the phytochemicals present in the food (Santos & Gonçalves 2016). The phenolic compounds of chia are polar, so they could be extracted easily with water. However, the aqueous extraction in Chia Seeds is not possible due to the release of mucilage.

#### DPPH RADICAL SCAVENGING ACTIVITY

Figure 2 shows the DPPH scavenging activity of the extracts with and without hexane treatment. With exception of 100% ethanol extract without treatment, the extracts gave a high DPPH radical scavenging activity (>80%). Statistically only significant differences were found in antioxidant capacity of the 100% ethanol extract without hexane treatment. This extract presented a lower radical scavenging capacity and total phenolic content. The treatment with hexane influenced on the antioxidant capacity of the 100% ethanol extract, which allowed that the activity increase. This may be attributable to the fact that hexane treatment can eliminate compounds of different nature that interfere with the antioxidant activity of phenolic compounds.

No significant difference was found between the DPPH radical scavenging activity of the methanol extracts hexane treated and not treated. Similarly, no significant difference was found between the antioxidant capacity of the 50-90% aqueous ethanol and methanol extracts hexane treated and not treated. There are no reports of the DPPH scavenging activities of the ethanol extracts to compare these results. Martínez-Cruz and Paredes-López (2014) in the 70% aqueous methanol extract of Chia Seeds reported an inhibition of DPPH of 68.8%. In this study, the DPPH radical scavenging activity of the extract obtained with the same concentration and conditions was higher ( $85.1 \pm 1.6\%$ ). Silveira and Salas (2014) reported a 50% reduction of the DPPH radical with the extract generated with 100% methanol. However, in this work, a greater DPPH radical scavenging activity was found ( $88.2 \pm 1.3\%$ ) in the extracts with the

same concentration of the methanol. The DPPH radical scavenging activity of the Lamiaceae family has already been described in species such as *Salvia officinalis* ( $92.3 \pm 0.5\%$ ), *Salvia sclarea* ( $92.9 \pm 0.4\%$ ), *Salvia glutinosa* ( $91.5 \pm 0.5\%$ ) and *Salvia pratensis* ( $93.0 \pm 0.5\%$  IDPPH) (Miliauskas et al. 2004).

A correlation analysis was performed between total phenol content and the DPPH radical scavenging activity of the extracts. The ethanol extracts with and without hexane treatment presented a positive relationship and mean correlation of  $R^2 = 0.67 \pm 0.07$  and  $R^2 = 0.67 \pm 0.12$ , respectively. The correlation in methanol extracts with and without hexane were lower than in ethanol extracts,  $R^2 = 0.40 \pm 0.2$ , and  $R^2 = 0.53 \pm 0.10$ . However, these

correlations were not statistically different according to the Tukey test ( $P < 0.05$ ). Some investigations found a strong correlation between the total content of phenolic compounds and the DPPH radical scavenging activity (Repo-De-Carrasco & Encina-Zelada 2008). However, other authors explain that these correlations are not completely linear, since it has been observed that the aqueous organic extracts may contain other non-antioxidant food constituents that may interfere in the assays of the antioxidant capacity and quantification of phenolic compounds, such as vitamins and minerals (MacDonald-Wicks et al. 2006; Santos & Gonçalves 2016).

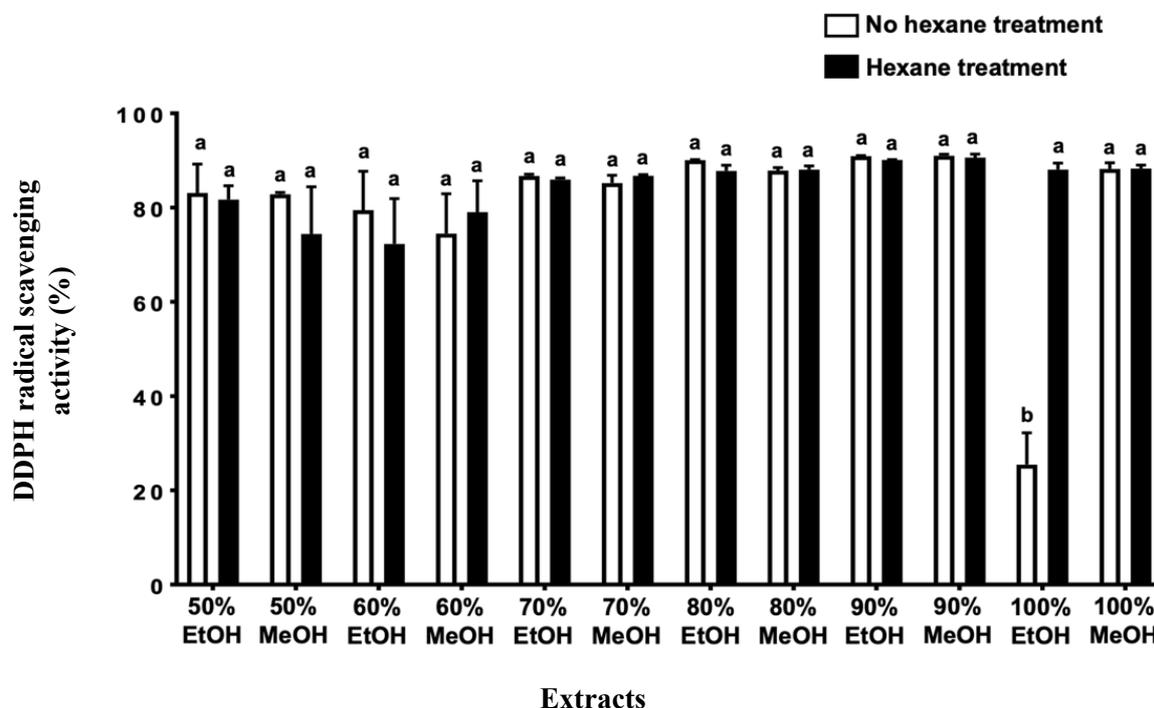


FIGURE 2. DPPH radical scavenging activity of ethanol (EtOH) and methanol (MeOH) extracts of *Salvia hispanica* L. seeds with and without hexane treatment. Columns with different letter are statistically different according to the Tukey test ( $P < 0.05$ )

#### IC<sub>50</sub> AND EXTRACTION YIELDS

Table 1 shows the IC<sub>50</sub> of the 80% aqueous alcohol extracts and their extraction yields. These samples were selected

because they showed a high DPPH radical scavenging activity and water solubility. The extracts were re-dissolved in water because the main application of antioxidant plant extracts is in the food manufacturing.

This industry restricts the use of certain solvents due to their toxicity. In addition, preliminary solubility tests were performed to choose the solvent. The extracts were re-dissolved in water and in the respective alcohols and

the content of phenolic compounds in the solutions was quantified. In these tests, no significant differences were found between the samples.

TABLE 1. IC<sub>50</sub> and extraction yield of the 80% aqueous ethanol and methanol extracts of the *Salvia hispanica* L. seeds

Extracts	IC <sub>50</sub> (mg/mL)	Extraction Yield (%)
80% aqueous ethanol	No hexane treatment	2.2 ± 0.1 <sup>a</sup>
	Hexane treatment	1.9 ± 0.2 <sup>a,b</sup>
80% aqueous methanol	No hexane treatment	1.5 ± 0.0 <sup>b</sup>
	Hexane treatment	1.8 ± 0.1 <sup>b</sup>

Means with different letters are significantly different ( $P < 0.05$ ) according to the Tukey test

The IC<sub>50</sub> of the 80% aqueous ethanol and methanol extracts were better than reported by Scapin et al. (2016) (IC<sub>50</sub> = 3.8 ± 0.1 mg/mL). They evaluated the antioxidant activity of the 80% aqueous ethanol extracts of Chia Seeds obtained under heating conditions (40 °C). Chia isolates have higher antioxidant capacity than mint (*Mentha arvensis*) (IC<sub>50</sub> = 17.4 mg/mL) (De Morais et al. 2009), Star fruit (*Averrhoa carambola*) (IC<sub>50</sub> = 3.8 ± 2.1 mg/mL) (Lim et al. 2006) and Brazilian berry (*Myrciaria jaboticaba*) (IC<sub>50</sub> = 3.54 mg/mL) (Da Silva et al. 2017). No significant differences were found between the IC<sub>50</sub> of extracts with and without hexane treatment. However, there are differences between the IC<sub>50</sub> of 80% aqueous ethanol and methanol extracts without hexane treatment. Among these, 80% aqueous ethanol extracts have a lower antioxidant capacity. In the literature there are no reports of the extraction yields of ethanol and methanol extracts in Chia Seed. The results obtained are similar to those reported in alcohol extracts of other seeds, such as *Hordeum vulgare* L. and *Hibiscus cannabinus* L. (Anwar et al. 2010; Yusri et al. 2012). Some authors describe that the extraction yields are dependent on the solvent used

(Goli et al. 2004; Stalikas 2007) but in this investigation no significant differences were found between the aqueous ethanol and methanol extracts yields.

#### SPECTRAL SCANNING BY UV-VISIBLE AND STRUCTURAL ANALYSIS BY FTIR

Figure 3 shows the UV-Visible spectral scanning of the 80% aqueous ethanol and methanol extracts and standards in the wavelength range of 200-500 nm. From 500 nm no absorbance of the samples was observed. In aqueous alcohol extracts of Chia Seed, flavonoid compounds such as myricetin, quercetin and kaempferol have been identified, as well as phenolic acids (chlorogenic, caffeic, and cinnamic) (Capitani et al. 2014; Marineli et al. 2014; Martínez-Cruz & Paredes-López 2014; Silveira & Salas 2014). The absorbance profile observed in 80% aqueous ethanol and methanol extracts of chia is compatible with flavonoids. These molecules present a maximum absorbance between 240-290 nm due to the absorption of A ring and another maximum greater than 300 nm depending of the substitutions and conjugations of the C ring (Zhao et al. 2006).

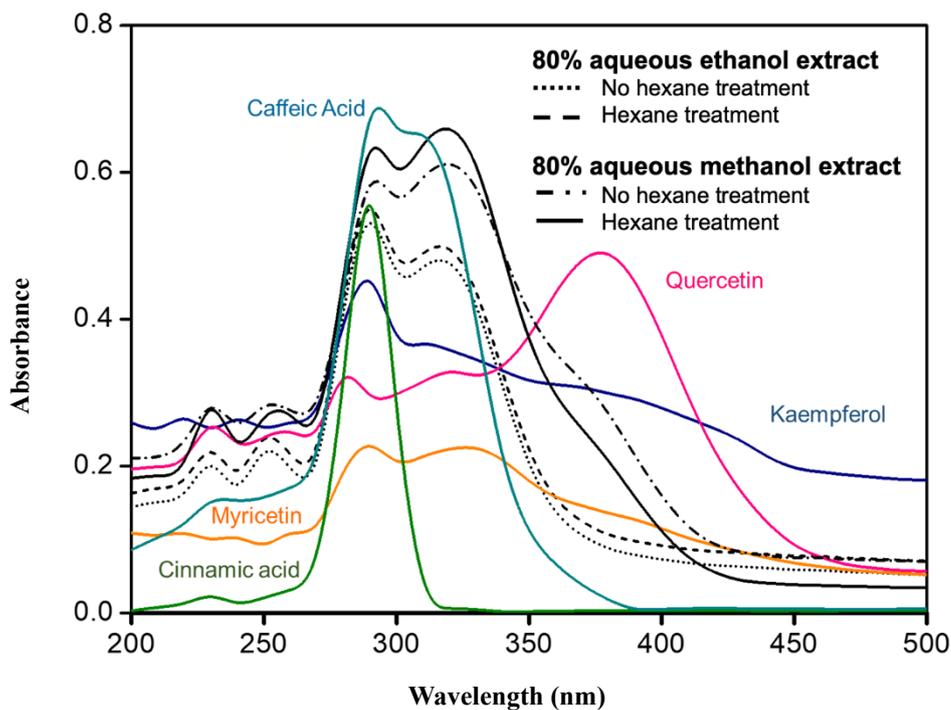


FIGURE 3. UV-Visible spectra of the 80% aqueous ethanol and methanol extracts of *Salvia hispanica* L. seeds with and without hexane treatment and standards (myricetin, quercetin, kaempferol, cinnamic and caffeic acid)

Differences were found between the UV-Visible spectra of the aqueous ethanol and methanol extracts, but not between the spectra of their respective ones without and with hexane treatment. The UV-Visible spectra of the

80% aqueous ethanol extracts showed peaks at 230, 250, 290, and 320 nm, obtaining the maximum absorbance at 290 nm. This peak is also presented in the spectrums of the kaempferol and the cinnamic acid. In addition, the

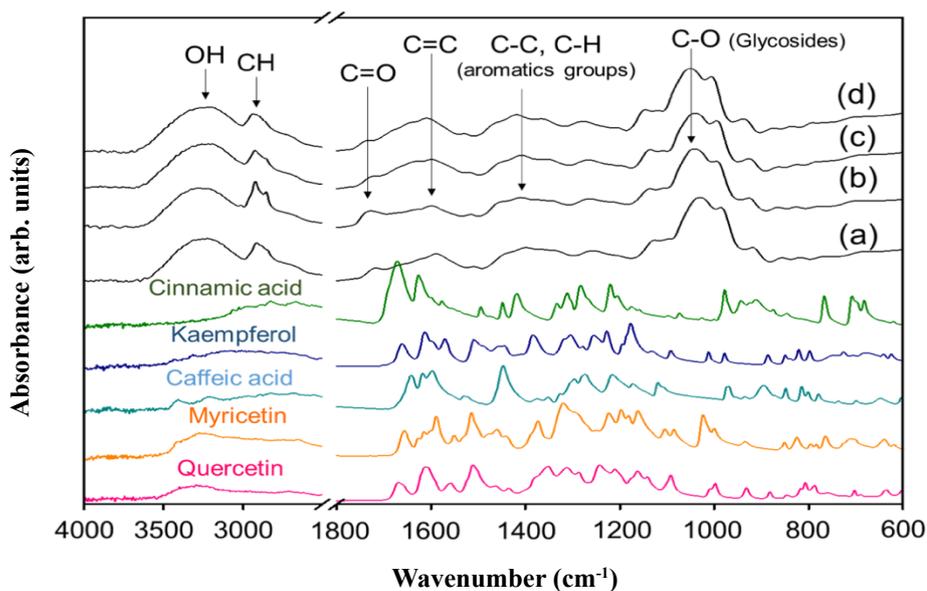


FIGURE 4. FTIR spectra of the 80% aqueous ethanol extract with hexane treatment (a) 80% aqueous ethanol extract without hexane treatment (b), 80% aqueous methanol extract with hexane treatment (c) 80% aqueous methanol extract without hexane treatment extracts (d) of *Salvia hispanica* L. seeds and standards (myricetin, quercetin, kaempferol, cinnamic and caffeic acid)

absorbances of the 80% aqueous ethanol extracts are similar to the spectrum of caffeic acid. This acid exhibits two main peaks at 290, and 310 nm. In contrast, in the 80% aqueous methanol extracts the scanning showed absorbance peaks at 230, 250, 290, 320 nm and a broad band at 380 nm. The maximum absorbance peak was at 320 nm. This absorbance profile resembles the spectra of quercetin and myricetin, which could indicate the presence of these types of compounds in the extracts.

The FTIR spectra of the 80% aqueous ethanol and methanol extracts and the standards are shown in Figure 4. In all the extracts were observed peaks of greater intensity in the spectral interval of 1200-1000  $\text{cm}^{-1}$ . These bands correspond to the stretching vibrations of the C-O bond of the structure of glycosides (Silva et al. 2014). The phenolic compounds extracted from plants are not found in free form, they bind to carbohydrates forming structures known as glycosides (Khoddami et al. 2013). These signals are not present in the spectra of the standards.

The signals located between 1408-1598  $\text{cm}^{-1}$  are associated with the stretching vibrations of aromatics groups (Cobaleda et al. 2017). These bands are observed in the extracts of chia and in the standards. At 1725  $\text{cm}^{-1}$  appear the signal of stretching vibration of the carbonyl group. In the standards, this peak appears at 1661  $\text{cm}^{-1}$ , in the myricetin, 1658  $\text{cm}^{-1}$ , in the kaempferol and 1668  $\text{cm}^{-1}$  in the quercetin. The band at 2920  $\text{cm}^{-1}$  corresponds to the stretching vibration of the C-H bond. These bands are observed in the extracts of chia and in the standards. Finally, in the region of 3300-3000  $\text{cm}^{-1}$ , the signals corresponding to the O-H bonds of the alcohols are observed, which like the aromatic groups, are part of the flavonoid molecule and the phenolic acids. Among the FTIR spectra of the aqueous ethanol and methanol extracts, slight differences were found, but not between their respective with hexane treatments. Confirming as described above, the treatment with hexane does not generate changes in the extracts. The differences observed between the 80% aqueous methanol and ethanol extracts appear in the region corresponding to the carbonyl group. The spectra of the 80% aqueous ethanol without treatment have a greater intensity and definition in the peak at 1725  $\text{cm}^{-1}$ . These results in conjunction with the variations observed in the UV-Visible spectra may suggest that there are differences in the composition of ethanol and methanol extracts.

#### CONCLUSION

The results show that the phenolic compounds in *Salvia hispanica* L. seeds can be extracted with different ethanol and methanol solutions. The highest total phenolic

content was quantified in the 50% aqueous ethanol and methanol extracts. However, all the extracts have a high DPPH scavenging activity. The 80% aqueous methanol extracts presented a better antioxidant capacity than the 80% aqueous ethanol extracts. The UV-Visible and FTIR spectra evidenced that there are differences in the composition of the ethanol and methanol extracts. With exception of the 100% ethanol extract, the hexane treatment does not influence on the total phenolic content and the DPPH scavenging activity of the extracts. This procedure is not necessary if high purity extracts are not required, so extraction is economized. The profile of the total phenolic content and the antioxidant capacity obtained with different concentrations of ethanol and methanol can serve for choosing the right conditions for the isolation of antioxidants in Chia Seeds according to the application of the extract.

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